

Electrocatalytic Oxidation of Tyrosine by Parallel Rate-Limiting Proton Transfer and Multisite Electron-Proton Transfer [*J. Am. Chem. Soc.* **2006**, 128, 11020–11021]. Christine J. Fecenko, Thomas J. Meyer,* and H. Holden Thorp*

Table 1. Rate and Equilibrium Constants for Added Sodium Dibasic Phosphate and Histidine^a

base	p <i>K</i> a	$K_A (M^{-1})$	$k_1 (s^{-1})$	$k_{-1} \; (M^{-1} \; s^{-1})$	$k_2 (M^{-1} s^{-1})$	$K_{A'} (M^{-1})$	$k_{\rm red}~({\rm s}^{-1})$
histidine phosphate (HPO ₄ ²⁻)	6.6 7.2	26.3 ± 0.3 30.0 ± 0.1	$\begin{array}{c} 1.4 \pm 0.1 \times 10^5 \\ 3.3 \pm 0.1 \times 10^5 \end{array}$	$2.9 \pm 0.1 \times 10^9$ $7.8 \pm 0.4 \times 10^9$	$1.4 \pm 0.1 \times 10^7 1.7 \pm 0.3 \times 10^7$	37.8 ± 0.2 22.2 ± 0.1	$6.9 \pm 0.3 \times 10^4 9.6 \pm 0.5 \times 10^4$

^a According to Scheme 1, at room temperature in 0.8 M NaCl.

Page 11021. We report here corrected values for k_{-1} for back proton transfer in the mechanism in Scheme 1. The initial values did not include the total concentration of tyrosine in the calculated rate constants. We also report a corrected value for $k_{\rm red}$. The corrected Table 1 is given here.

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